

Dy-Fe-Pt (Dysprosium-Iron-Platinum)

V. Raghavan

Recently, [2007Lei] determined an isothermal section for this system at 900 °C, which depicts no ternary compounds.

Binary Systems

The Dy-Fe phase diagram [Massalski2] shows the following intermetallic compounds: Fe₁₇Dy₂ (Ni₁₇Th₂-type hexagonal), Fe₂₃Dy₆ (D8_a, Mn₂₃Th₆-type cubic), Fe₃Dy (Be₃Nb-type rhombohedral), and Fe₂Dy (C15, MgCu₂-type cubic). The Dy-Pt [2006Oka, 2007Lei] depicts the following compounds: Dy₃Pt (D0₁₁, Fe₃C-type orthorhombic), Dy₂Pt (C23, Co₂Si-type orthorhombic), Dy₅Pt₃ (D8₈, Mn₅Si₃-type hexagonal), Dy₅Pt₄ (Ge₄Sm₅-type orthorhombic), DyPt (B27, FeB-type orthorhombic), Dy₃Pt₄ (Pd₄Pu₃-type rhombohedral), DyPt₂ (C15, MgCu₂-type cubic), DyPt₃ (L1₂, AuCu₃-type cubic), and DyPt₅ (orthorhombic). The Fe-Pt phase diagram [Massalski2] shows three ordered phases: Fe₃Pt (L1₂, AuCu₃-type cubic), FePt (L1₀, AuCu-type tetragonal), and FePt₃ (AuCu₃-type cubic).

Ternary Isothermal Section

With starting metals of >99.9 mass % purity, [2007Lei] arc-melted under Ar atm 62 alloy samples containing up to 75 at.% Dy. The samples were annealed at 900 °C for 15 days and quenched in water. The phase equilibria were studied with x-ray powder diffraction, scanning electron microscopy, and energy dispersive spectroscopy. The isothermal section at 900 °C constructed by [2007Lei] is redrawn in Fig. 1. The solubility of Dy in (αFe), (γFe), FePt, FePt₃, and (Pt) was 3, 2, 2, 1.5, and 1.5 at.%, respectively. Fe₂Dy dissolves up to 5 at.% Pt. The solubility of Fe in Dy-Pt compounds was found to be less than 1 at.%. No ternary compounds were found.

References

- 2006Oka:** H. Okamoto, Dy-Pt (Dysprosium-Platinum), *J. Phase Equilib. Diffus.*, 2006, **27**(4), p 429
2007Lei: M. Lei, G. Zhengfei, Z. Xiaping, C. Gang, Z. Bo, and X. Chengfu, Solid State Phase Equilibria in the Fe-Pt-Dy Ternary System at 900 °C, *J Alloys Compd.*, 2007, **427**, p 130-133

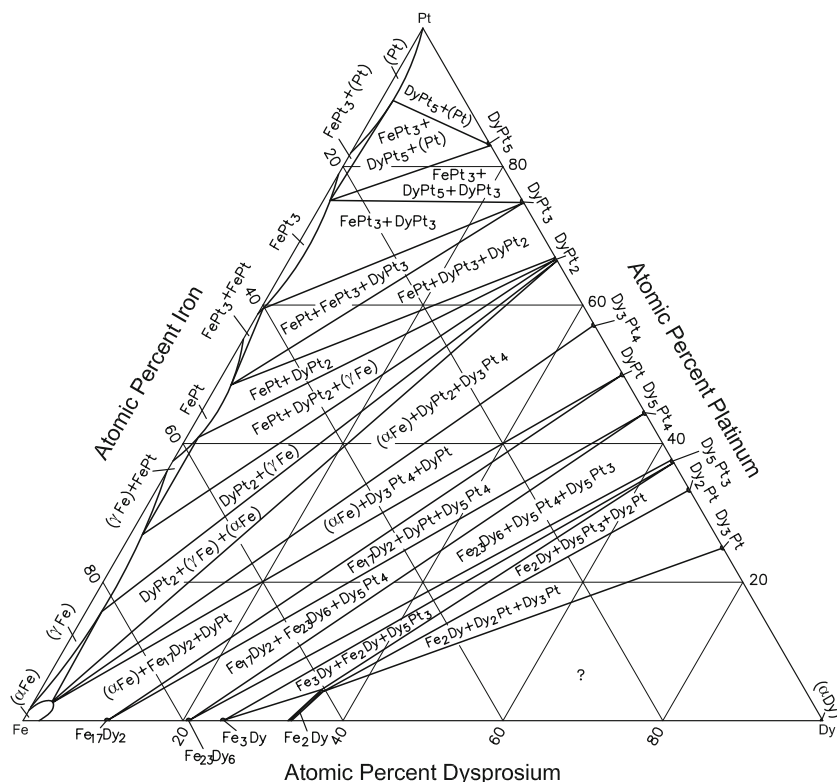


Fig. 1 Dy-Fe-Pt isothermal section at 900 °C [2007Lei]. Narrow two-phase regions are omitted